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**Question 11: What is the minimum hydrogen sulfide required in the recycle gas for units with low sulfur feed? Do you inject sulfur compounds to maintain a minimum concentration?**

**Kaspar Vogt (Albemarle)**

In some processes, hydrotreating catalysts are used to treat feedstocks containing very low sulfur (below 20 ppm). These processes can include the following:

1. Gulf HPG process for treating pyrolysis naphtha (second stage).
2. Two step naphtha hydrotreating process in steam reforming (ammonia synthesis process).
3. Treating olefinic fuel gas prior to reforming to make synthesis gas for methanol manufacturing.
4. Wax and certain lube hydrofining operations

It appears that in these processes the catalyst slowly loses its HDS activity. This is due to the transformation of the Molybdenum present as activator, and Nickel/Cobalt present as promoters, from their active sulfide form to the inactive metal state. This process is caused by a hydrogen sulfide partial pressure that is too low (see phase diagrams below).

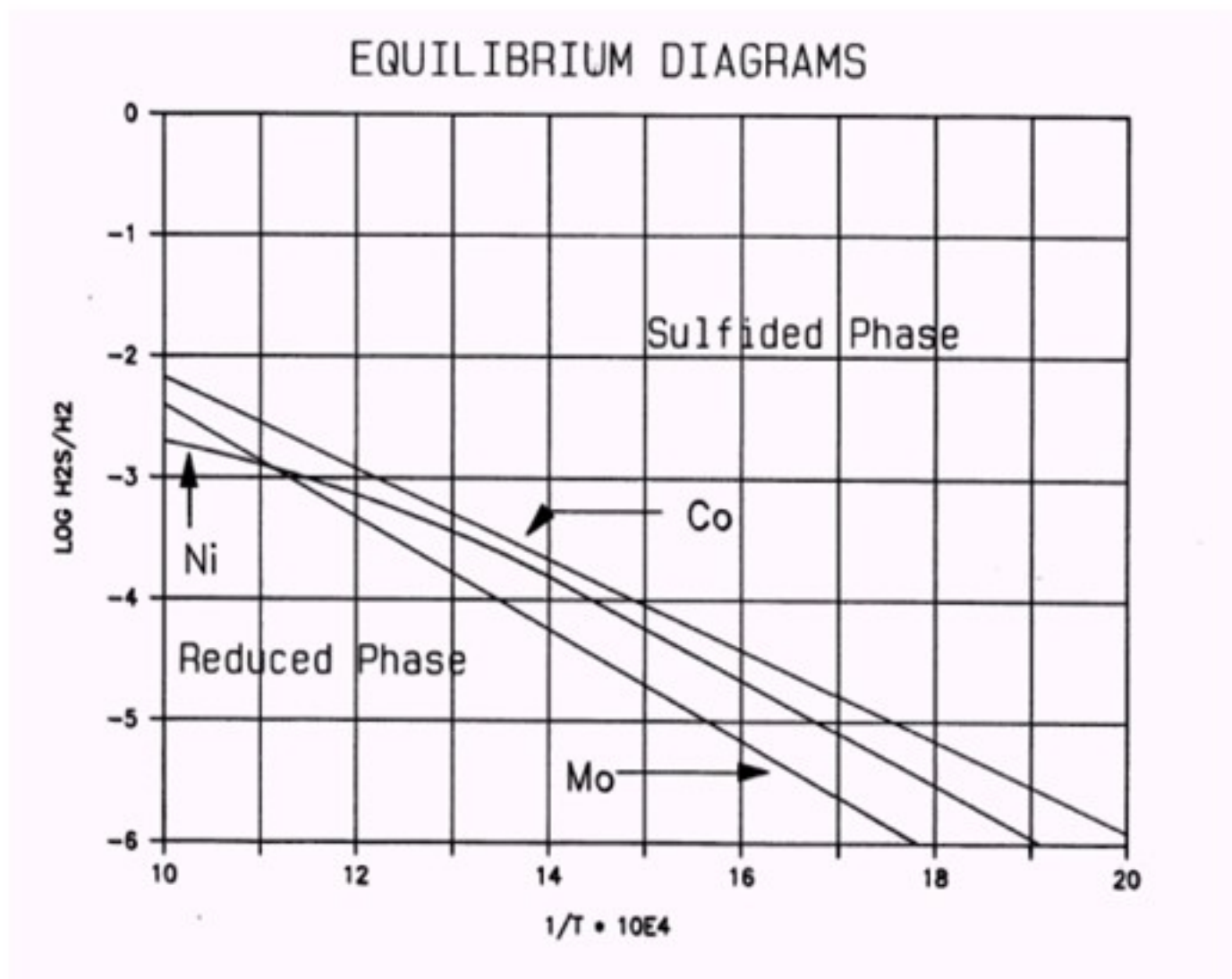


Figure 1. Equilibrium Diagram

In some cases (usually naphtha units) conditions can arise that contribute to a low H<sub>2</sub>S partial pressure and could result in metal reduction. Based on the feed properties, operating conditions and product objectives, we can determine whether the unit is or will be operating in a critical operating window. If this is the case, a spiking agent such as DMDS should be added to the feed to boost the H<sub>2</sub>S partial pressure.

From the phase diagram shown above, we see that the Cobalt is the most critical element. We can begin to move into the CoS<sub>9</sub>-phase by decreasing the temperature or increasing the H<sub>2</sub>S partial pressure by adding sulfur. To move into the "safe" region of the diagram, the Log<sub>10</sub> H<sub>2</sub>S/H<sub>2</sub> should be at least -4 at a temperature of 640 deg F.

If the same calculation is done in the reverse order the minimum feed sulfur to avoid metal reduction can be calculated. If we process a 1 ppm sulfur, 70°API naphtha feed at 360 psig at 640 deg F, 80% H<sub>2</sub> purity, down to 0.4 ppm sulfur in the product (actual sulfur removal is 0.6 ppm) we conclude that 20 ppm sulfur should be added to avoid catalyst metal reduction. Spiking can be done with a sulfiding agent,

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e.g., DMDS, DMS.

### **Minh Dimas (CITGO)**

For diesel hydrotreaters, as long as the material being fed to the reactor contains sulfur (i.e., sour feed) and the recycle gas contains 25-50 ppm H<sub>2</sub>S (by adjusting the amine circulation when necessary), there should not be a need to inject a sulfur-spiking agent into the feed. This ensures a small concentration of H<sub>2</sub>S at the inlet of the reactor. With reactor temperatures above catalyst activation temperatures, additional H<sub>2</sub>S is being generated from that point on through the reactor. For us, the sweetest feed is observed when reprocessing off-spec diesel, in which case we may shut down the Recycle H<sub>2</sub> Amine Scrubber to preserve the H<sub>2</sub>S and protect the catalyst.

### **Tim Lewer (Shell)**

In hydrotreating units where the feed sulfur is low and the temperature and H<sub>2</sub> partial pressure are low, operation with as low as 20 ppm H<sub>2</sub>S in the recycle gas has been observed. There is not, however, a universal minimum concentration requirement. The minimum required H<sub>2</sub>S concentration will vary from unit to unit depending on many factors including, but not limited to: unit feed sulfur concentration, unit pressure, reactor temperature, catalyst type, vent rate, catalyst age, and H<sub>2</sub> partial pressure. It is common practice to inject a sulfiding agent such as DMDS to maintain adequate H<sub>2</sub>S in the recycle gas. In addition, refiners have used sour make up gas streams to provide adequate H<sub>2</sub>S partial pressure. The guidelines can change based on what catalyst company you talk with, but it all depends on how conservative you want to be. You need to set the H<sub>2</sub>S low limit to provide a proper buffer zone. It is recommended to discuss H<sub>2</sub>S concentration requirements for all situations with your catalyst vendor to make sure the catalyst is properly protected against metal reduction.

### **Gordon Chu (ART)**

There is no minimum hydrogen sulfide requirement as long as the feed contains some sulfur as the sulfided catalyst is very resistant to sulfur loss under normal process conditions. We are not aware of any refiners adding sulfur compounds to maintain a minimum H<sub>2</sub>S concentration during the process cycle.

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